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# MICROSTRUCTURALLY MODIFIED ZINC CONTAINING PHOSPHOSILICATE BIOGLASSES APPLICABLE IN COMPOSITE SCAFFOLDS FOR TISSUE REGENERATION

ROZALIA VERES<sup>1,2</sup>, MONICA TĂMĂȘAN<sup>1</sup>, CONSTANTIN CIUCE<sup>2</sup>,  
VIORICA SIMON<sup>1</sup>

<sup>1</sup>Babeș-Bolyai University, Faculty of Physics & Institute of Interdisciplinary Research in Bio-Nano-Sciences, Cluj-Napoca

<sup>2</sup>Iuliu Hațieganu University of Medicine and Pharmacy, Faculty of Medicine, Cluj-Napoca

## Abstract

**Aims.** Bioactive glasses have attracted significant attention in recent years, especially for their insertion in composite biomaterials to serve as support in tissue engineering. These biomaterials have clinical applications also as a coating for implants, bone filling and, in a porous form they are used for bone regeneration. Different bioactive glasses and glass-ceramics have been synthesized in order to get desired properties conferred by required microstructure. The aim of this work is to obtain and characterize the structure and surface of a new bioglasses in zinc containing phosphosilicate system, applicable as scaffold for tissue regeneration.

**Methods.** The new  $\text{SiO}_2\text{-ZnO-P}_2\text{O}_5$  bioglass system, with four different molar ratios, was prepared following the sol-gel route. The precursors used were TEOS ( $\text{Si}(\text{OC}_2\text{H}_5)_4$ ), zinc nitrate hexahydrate ( $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ) and dibasic ammonium phosphate ( $(\text{NH}_4)_2\text{HPO}_4$ ), while  $\text{HNO}_3$  was used as catalyst. The obtained gels were dried for 24 hours at 110 °C. For the characterization of samples was used differential thermal analysis (TGA / DTA), X-ray diffraction (XRD) as well as atomic force microscopy (AFM).

**Results.** According to thermal analysis results, heat treatments at 300 °C and 750 °C are necessary for removal of water and residual groups from the investigated samples. XRD data show that all samples are prevalent amorphous, but they also contain nanometric crystals of zinc silicates and phosphates developed during sol-gel synthesis. After the heat treatments performed at 300 °C and 750 °C, the XRD patterns prove that the samples are still predominantly amorphous. AFM analysis of samples shows a rough surface with high specific area.

**Conclusions.** The structure of the new  $\text{ZnO-P}_2\text{O}_5\text{-SiO}_2$  bioglass processed by sol-gel route remains predominantly amorphous after the heat treatments applied at 300 °C and 750 °C, and in the prevalent vitreous matrix are developed nanosized crystallites. AFM analysis reveals a rough surface with high surface area. Both the nanostructure and the large surface area of these samples are promising for the attachment of biomolecules.

**Keywords:** biomaterials, sol-gel, scaffold, bone.

## BIOSTICLE FOSFOSILICATICE CU ZINC MODIFICATE MICRO-STRUCTURAL, APLICABILE CA SUPTOR PENTRU REGENERARE TISULARĂ

### Rezumat

**Scop.** Sticlele bioactive au atras o atenție semnificativă în ultimii ani, mai ales pentru introducerea lor în biomaterialele compozite servind ca suport în ingineria tisulară. Aceste biomateriale au aplicații clinice ca înveliș pentru implanturi, în umplere osoasă și, într-o formă poroasă, acestea sunt folosite pentru regenerarea

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osoasă.

Scopul lucrării este obținerea și caracterizarea structurii și suprafeței unor noi biosticle într-un sistem fosfosilicatic cu zinc, aplicabile în biomateriale compozite ca suport pentru refacerea țesutului osos.

**Metode.** Noul sistem de biosticlă  $\text{SiO}_2\text{-ZnO-P}_2\text{O}_5$ , cu patru raporturi molare diferite, a fost preparat urmând tehnica sol-gel. Precursorii folosiți au fost TEOS ( $\text{Si}(\text{OC}_2\text{H}_5)_4$ ), azotat de zinc hexahidratat ( $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ) și fosfat de amoniu bibazic ( $(\text{NH}_4)_2\text{HPO}_4$ ), folosind  $\text{HNO}_3$  ca și catalizator. Gelul obținut a fost uscat 24 ore la  $110^\circ\text{C}$ . Pentru caracterizarea probelor s-a folosit analiza termică diferențială (TGA/DTA), difracția de raze X (XRD), precum și microscopia de forță atomică (AFM).

**Rezultate.** Din analiza termică a probelor rezultă că, pentru eliminarea apei și a elementelor reziduale, sunt necesare tratamente termice la  $300^\circ\text{C}$  și  $750^\circ\text{C}$ . Rezultatele XRD au arătat că toate probele uscate la  $110^\circ\text{C}$  au un caracter predominant amorf și apar cristale nanometrice de silicați și fosfați de zinc formați în timpul sintezei probelor. În urma tratamentelor efectuate la  $300^\circ\text{C}$  și  $750^\circ\text{C}$ , din difractograme reiese faptul că probele sunt încă predominant amorse. Analiza AFM a probelor evidențiază o suprafață rugoasă, cu arie specifică ridicată.

**Concluzii.** Noua biosticlă  $\text{SiO}_2\text{-ZnO-P}_2\text{O}_5$  obținută prin metoda sol-gel, tratată termic la  $300^\circ\text{C}$  și  $750^\circ\text{C}$  are o structură predominant amorfă, în care se formează cristalite nanometrice. Analiza AFM pune în evidență o suprafață rugoasă. Atât nanostructurarea probelor, cât și suprafața specifică mare sunt caracteristici urmărite în vederea atașării de biomolecule.

**Cuvinte cheie:** biomateriale, sol-gel, suport, os.

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## INTRODUCTION

Bones are very important in our lives because they support our bodies and enable us to perform various motions. When the bone is damaged and if an area of damaged bone is too large for self-repair, the damaged bones must be repaired by using alternative materials, such as autografts, allografts and artificial materials [1]. Autografts, which are transferred from healthy parts of the bones of the same patient, are widely used because they show high performance. Although allografts, which are transferred from other people, are also used, they have problems related to foreign body reactions and infections. However, sufficient material cannot always be derived from the patient or a suitable donor and current medical technologies are not yet able to synthesize these materials. As a consequence, artificial materials are substituted. Initially, these materials implanted in bony defects were encapsulated by fibrous tissue and do not bond to living bone. To solve the problem of the foreign body reaction, bioactive glasses have received much attention. For repairing bony tissues, the bioactivity is regarded as the capability to make direct contact with living bone after implantation in bone defects [2-3]. Different bioactive glasses and glass-ceramics have been synthesized in order to get desired properties conferred by required microstructure. The properties that are of interest in the characterization of biomaterial surfaces include the chemical composition, the hydrophilicity or hydrophobicity, the presence of

ionic groups, the morphology (i.e. the domain structure), and the topography (i.e. the surface roughness, planarity, and feature dimensions) that can influence the attachment of bio-organic molecules such as enzymes, proteins, antibodies or even cells [4-6]. Common components used for synthesis of 45S5 and S53P4 registered bioglasses are  $\text{Na}_2\text{O}$ ,  $\text{CaO}$ ,  $\text{P}_2\text{O}_5$ ,  $\text{SiO}_2$  [7]. In addition to these are often added  $\text{K}_2\text{O}$ ,  $\text{MgO}$ ,  $\text{B}_2\text{O}_3$ . There are some other glasses and glass-ceramics which also include  $\text{ZnO}$ ,  $\text{Ag}_2\text{O}$  and  $\text{Al}_2\text{O}_3$ . A major disadvantage of bioglass is its very high solubility because most of the released ions might be transported away from the surroundings of the implantation site by body fluid before new bone can form. Control of the surface reactions and, therefore, of the biodegradation and bioactivity of implanted materials can be achieved by rational design of the glass composition. In fact, by varying the chemical nature and/or concentration of the bioglass constituents, new important mechanical and biological properties can be tailored for specific clinical applications [8]. The bioactivity and the bioresorption of ceramics is governed by the reaction between the ceramic and the body fluid. Zinc oxide addition to the phosphosilicate matrix was considered because it is known that zinc stimulates cell proliferation, its stimulating effect being demonstrated *in vitro* [9]. In addition, Zn slows the glass dissolution. From the literature it is known that  $\text{ZnO-P}_2\text{O}_5\text{-SiO}_2$  system has not been studied until now.

Techniques for preparing bioactive glasses include both traditional melting methods and sol-gel route, the latter being covered in this paper [10]. By comparison with

typical melt-derived bioactive glasses, sol-gel bioactive glasses have a larger surface area and OH groups are present in their structure. Therefore, sol-derived bioglasses show higher bioactivity than melted ones by faster formation of a strong bond with the bone tissue as well by the stimulation of living bone to faster regeneration [11]. Other advantages of the sol-gel process are well known: it takes place at low temperatures, and gives homogeneous mixtures in the final glass composition.

The aim of this work is to obtain and characterize the structure and surface of a new bioglass in zinc containing phosphosilicate system, applicable as scaffold for tissue regeneration.

## METHODS

The new  $\text{SiO}_2\text{-ZnO-P}_2\text{O}_5$  bioglass system, with four different molar ratios (table no I.), was prepared following the sol-gel route.

**Table no I.** Molar ratios of sol-gel derived  $\text{SiO}_2\text{-ZnO-P}_2\text{O}_5$  bioglasses.

Sample code	$\text{SiO}_2\text{:ZnO:P}_2\text{O}_5$ ratio	$\text{SiO}_2$ (mol %)	ZnO (mol %)	$\text{P}_2\text{O}_5$ (mol %)
S-414	4:1:4	44.5	11	44.5
S-412	4:1:2	57	14	29
S-411	4:1:1	66	17	17
S-4103	4:1:0.3	75	19	6

The precursors used were TEOS ( $\text{Si}(\text{OC}_2\text{H}_5)_4$ ), zinc nitrate hexahydrate ( $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ) and dibasic ammonium phosphate ( $(\text{NH}_4)_2\text{HPO}_4$ ). The catalyst used was  $\text{HNO}_3$ . TEOS was stirred with  $\text{C}_2\text{H}_5\text{OH}$ , the other two precursors were stirred separately with distilled water for 1/2 h. Then both solutions were added to TEOS solution under continuous stirring. Solution was left overnight for gelation at the room temperature. The gel was kept for 7 days at 37 °C for aging and then was dried for 24 h at 100 °C.

Differential thermal analysis (DTA) and thermogravimetric analysis (TGA) were performed on Shimadzu type derivatograph DTG-GOH with a heating rates of 5 °C/min using alumina open crucibles, in order to investigate the thermal behavior of the dried samples.

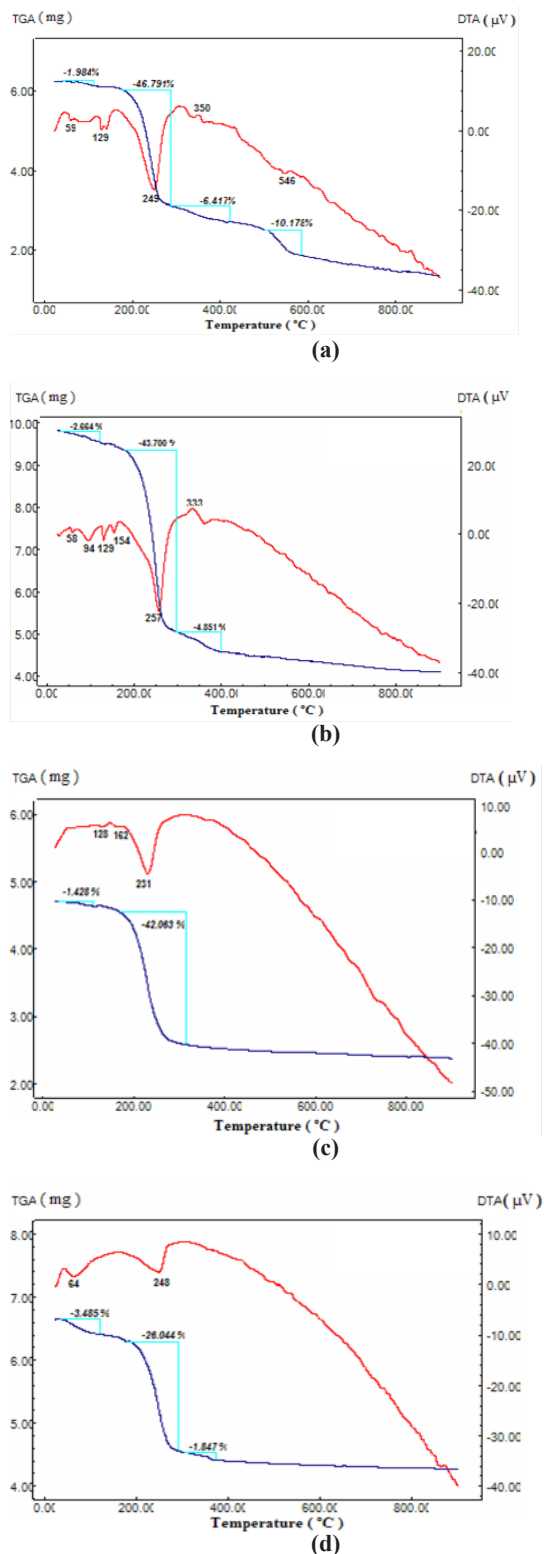
After DTA/TGA analysis, the samples were heat treated at 300 °C and 750 °C.

The structure of the dried and heat treated samples were examined using a X-ray Shimadzu XRD6000 with a Ni – filter, using  $\text{CuK}_\alpha$  radiation ( $\lambda = 1.5418\text{\AA}$ ). The operation voltage and current were 40 kV and 30 mA. The measurements were performed at a scan speed of 1°/min in a 2θ scan range from 10° to 80°.

Topographic study of samples pellets for  $\text{SiO}_2\text{-ZnO-P}_2\text{O}_5$  system was performed with an NT-MDT atomic force microscope in tapping-mode. Bi-dimensional topographic images with 10μm x 10 μm area were obtained and analyzed three-dimensional images with a resolution of 256 pixels.

## RESULTS

The TGA/DTA curves of 110 °C dried samples are shown in figure no 1.



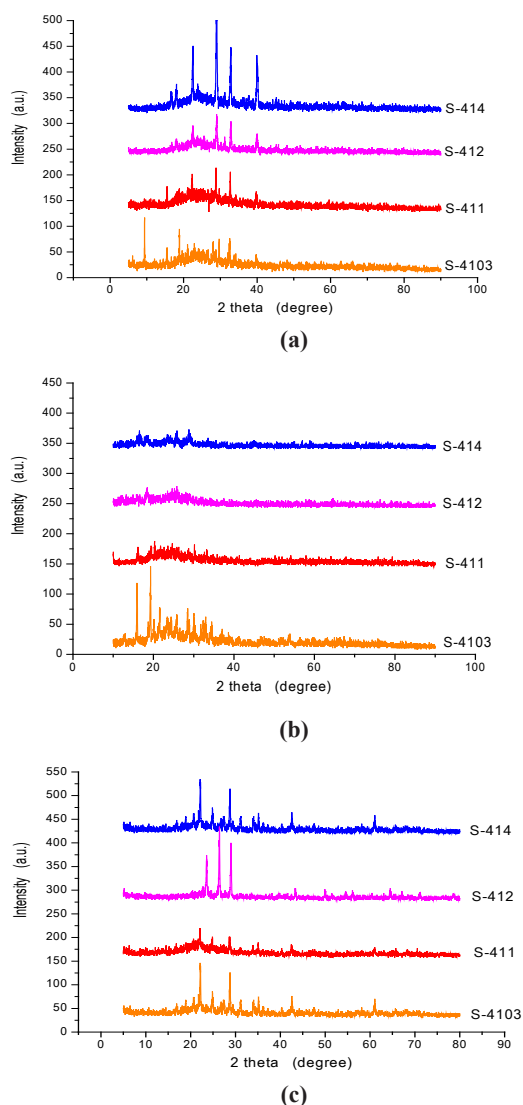
**Fig. 1.** DTA and TGA runs of sol-gel derived S-414 (a), S-412 (b), S-411 (c), S-4103 (d) bioglasses.

The TGA/DTA events temperature and weight losses are presented in table no.II.

**Table no II.** Temperature and weight losses related to DTA and TGA events in sol-gel derived  $\text{SiO}_2\text{-ZnO-P}_2\text{O}_5$  bioglass.

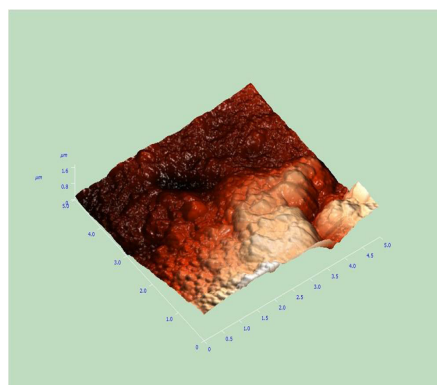
Sample code	Endothermic events temperature (°C)	Weight losses (%)
S-414	59, 129, 249, 546	1.984, 46.791, 6.417, 10.178
S-412	58, 94, 129, 154, 275	2.664, 43.700, 4.851
S-411	128, 162, 231	1.428, 42.063
S-4103	64, 248	3.485, 26.044, 1.847

The XRD patterns of the 110 °C dried bioglass sample and 300 and 750 °C heat treated samples are illustrated in figure no. 2.

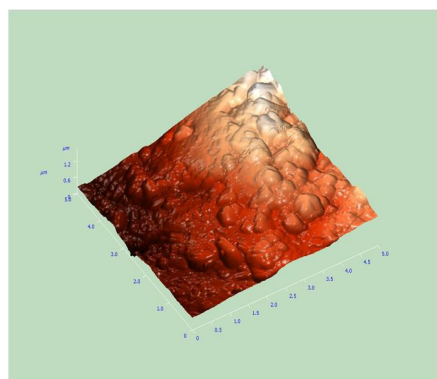


**Fig. 2.** XRD pattern of 110 °C dried (a), 300 °C (b) and 750 °C (c) heat treated sol-gel samples.

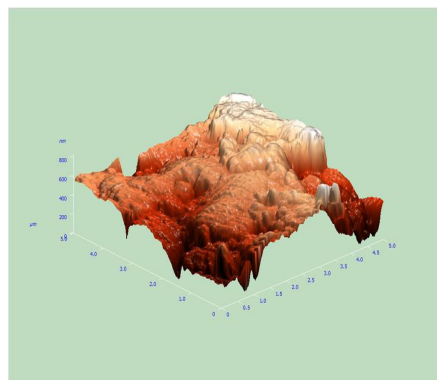
The three-dimensional (3D) views of the surface morphology of 750 °C heat treated samples pellets are shown in figure no 3.



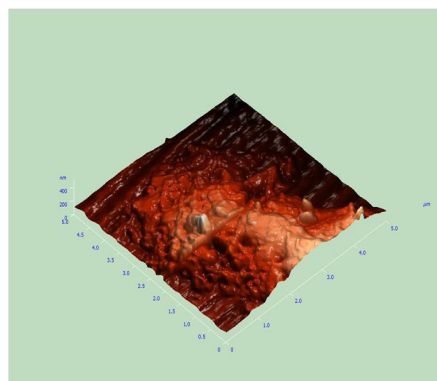
(a)



(b)



(c)



(d)

**Fig. 3.** AFM tapping mode 3D image of 750 °C heat treated S-414 (a), S-412 (b), S-411 (c), S-4103 (d) samples.

## DISCUSSIONS

The TGA curves for all the samples present more regions of weight loss. The first weight loss that occurs coincides with an endothermic peak in DTA signal and it can be associated with the remove of free water molecules. A second big weight loss with a large corresponding endothermic peak in the DTA curve can be observed around 250 °C. This loss can be associated with the decomposition of ammonium nitrate and the elimination of water caged in the pores. The others weight losses are small and the DTA signal is diminished constantly and can correspond to the elimination of the residual elements from precursors resulted during the synthesis of samples. Thermal analysis indicates that the elimination of the residual elements take place up to around 750 °C temperatures, which can be considered as thermal treatment temperature.

The XRD patterns of the 110 °C dried samples indicate a prevalent amorphous state of all glasses. Beside the large diffraction peak that dominates the XRD patterns, there are peaks indicating the presence of a minor crystalline structure, that could be assigned to residual zinc nitrate and phosphate crystals zinc silicates and phosphates crystals formed during the samples synthesis [12,13]. After thermal treatment at 300 °C it can observe of no peaks in the patterns that is indicative of their amorphous nature, except S-4103 that still has nano-crystals. The XRD results of 750 °C heat treated samples indicate that the samples are still amorphous, except S-412.

Surface morphology is well revealed in AFM images. Their analysis denotes a rough surface with high specific area [13].

## CONCLUSIONS

The structure of the new  $\text{ZnO-P}_2\text{O}_5\text{-SiO}_2$  bioglass processed by sol-gel route remains predominantly amorphous after the heat treatments applied at 300 °C and 750 °C, and in the prevalent vitreous matrix are developed nanosized crystallites. AFM analysis reveals a rough surface with high surface area. Both the nanostructure and the large surface area of these samples are promising for the attachment of biomolecules.

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